

REDUCTIVE BIOLOGICAL TREATMENT OF TEXTILE EFFLUENTS

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The beginning

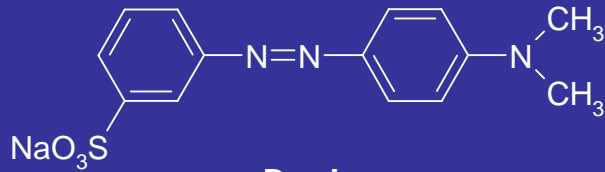
- Collection of contaminated soil from a treatment plant that receives large amounts of coloured water
- Growth in rich medium to select microorganisms
- Isolation of yeasts in selective medium
- Growth of yeasts in YEPD-agar medium with a model azo dye to select strains with decolourising capability
- Identification of the yeasts with better decolourising activity: *Candida zeylanoides* (UM2) and *Issatchenkia occidentalis* (UM41)



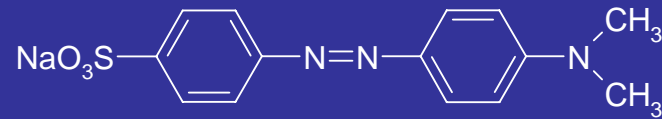
Study the decolourisation process by YEASTS

The decolourisation process by yeasts

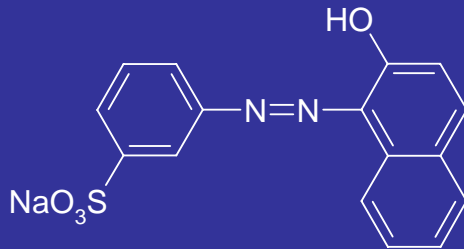
Selection of model azo dyes



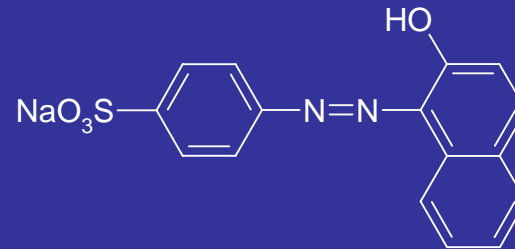
Dye I



Dye II (Methyl Orange)



Dye III



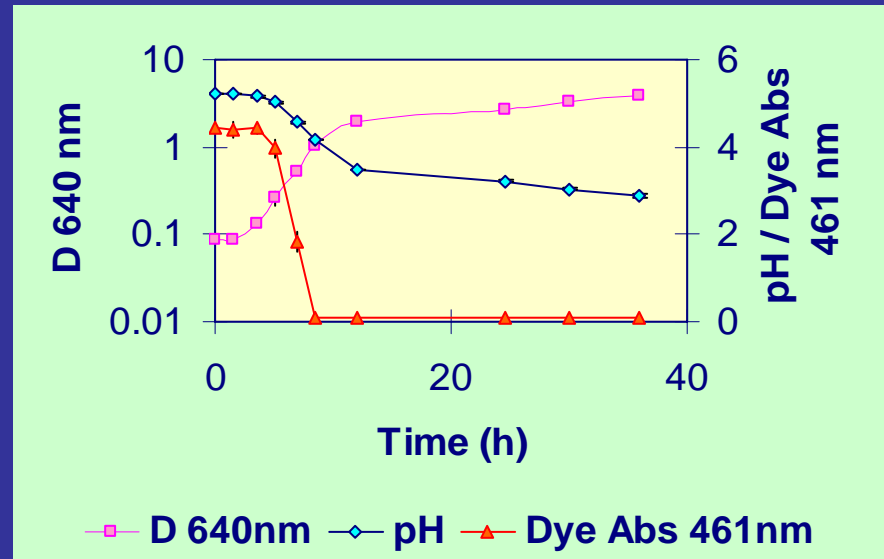
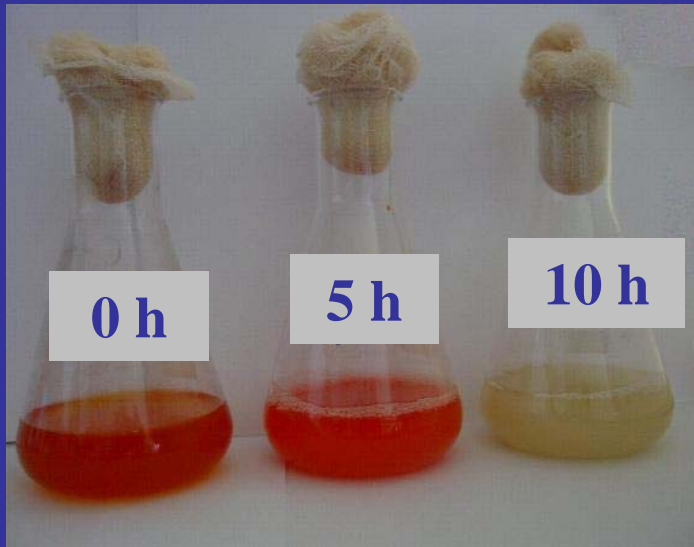
Dye IV (Orange II)



Amaranth

The decolourisation process by yeasts

Decolourisation in batch:



Conditions:

- 0.2 mM dye
- NDM with 2% glucose
- 120 rpm
- 26°C

The decolourisation process by yeasts

Decolourisation in batch - results:

- The presence of dyes in the growth medium does not affect the specific growth rates;
- Yeasts are able to reduce azo dyes to colourless amines;
- This activity is constitutive;
- Depends on dye structure.

The decolourisation process by yeasts

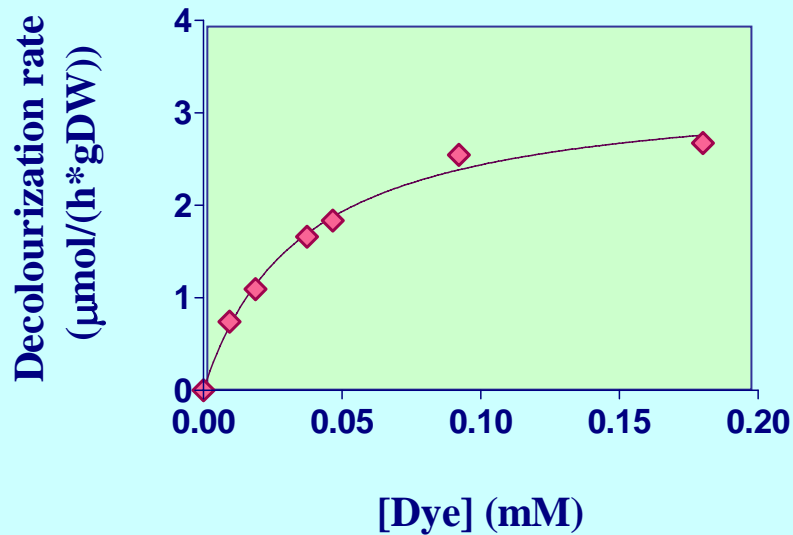
Decolourisation in batch - results:

- Dye degradation is connected to growth phase;
- The glucose is fermented and the resulting ethanol is respired;
- O₂ in the incubation conditions is limited;
- Under anoxic or aeration conditions there is no decolourisation;
- Neither dyes nor resulting amines are toxic to yeasts;
- Yeasts can use the produced amines as carbon and nitrogen sources.

The decolourisation process by yeasts

Activity assays - results:

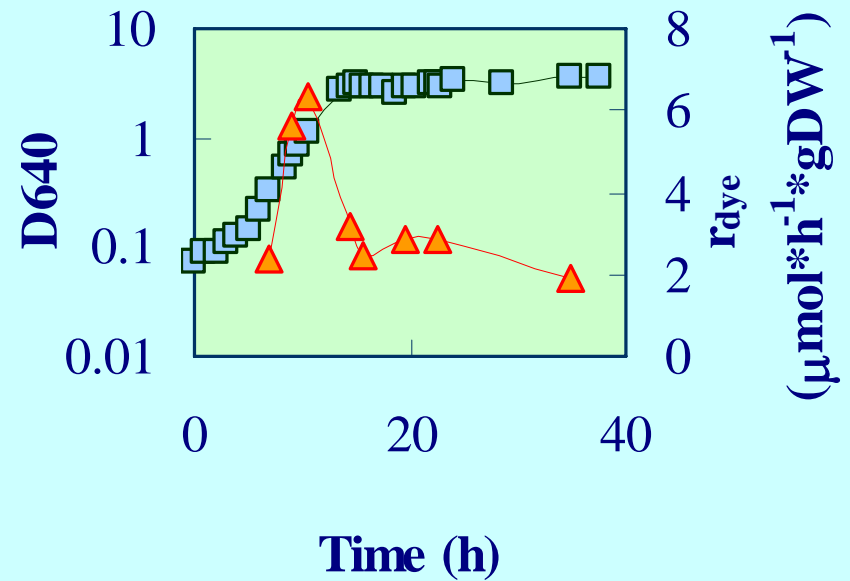
Effect of dye concentration



$$r_{\text{dye max}} = 3.2 \mu\text{mol} \cdot \text{h}^{-1} \cdot \text{g}^{-1}$$

$$k_M = 0.034 \text{ mM}$$

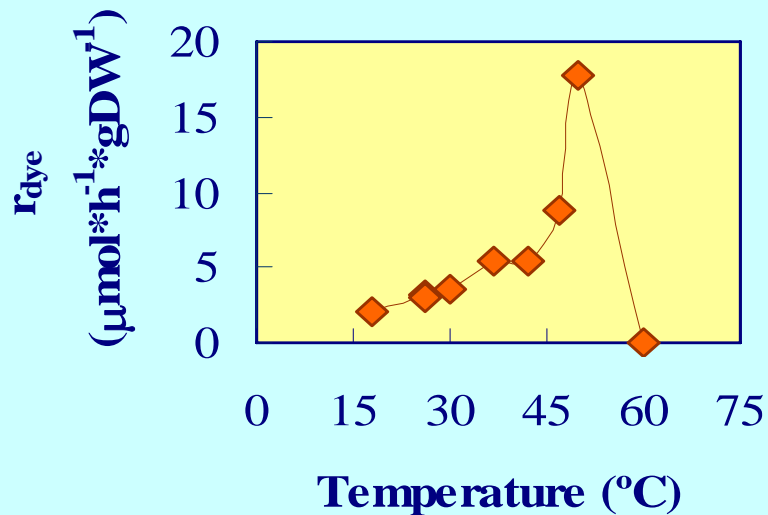
Effect of growth phase



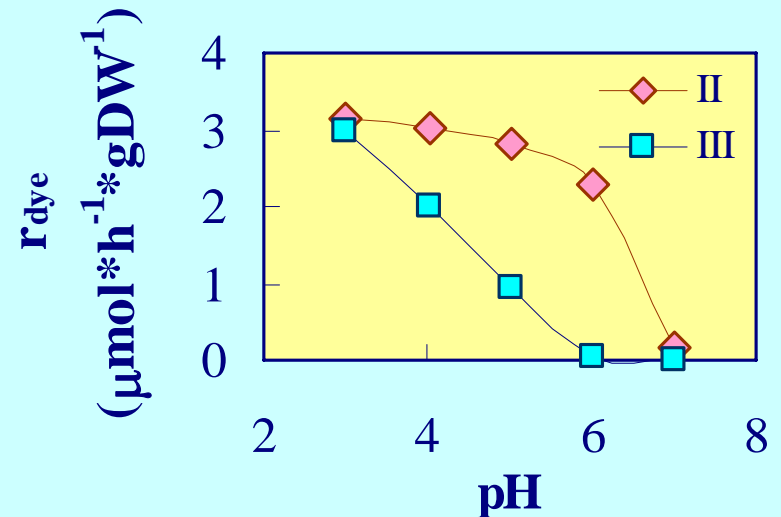
The decolourisation process by yeasts

Activity assays - results:

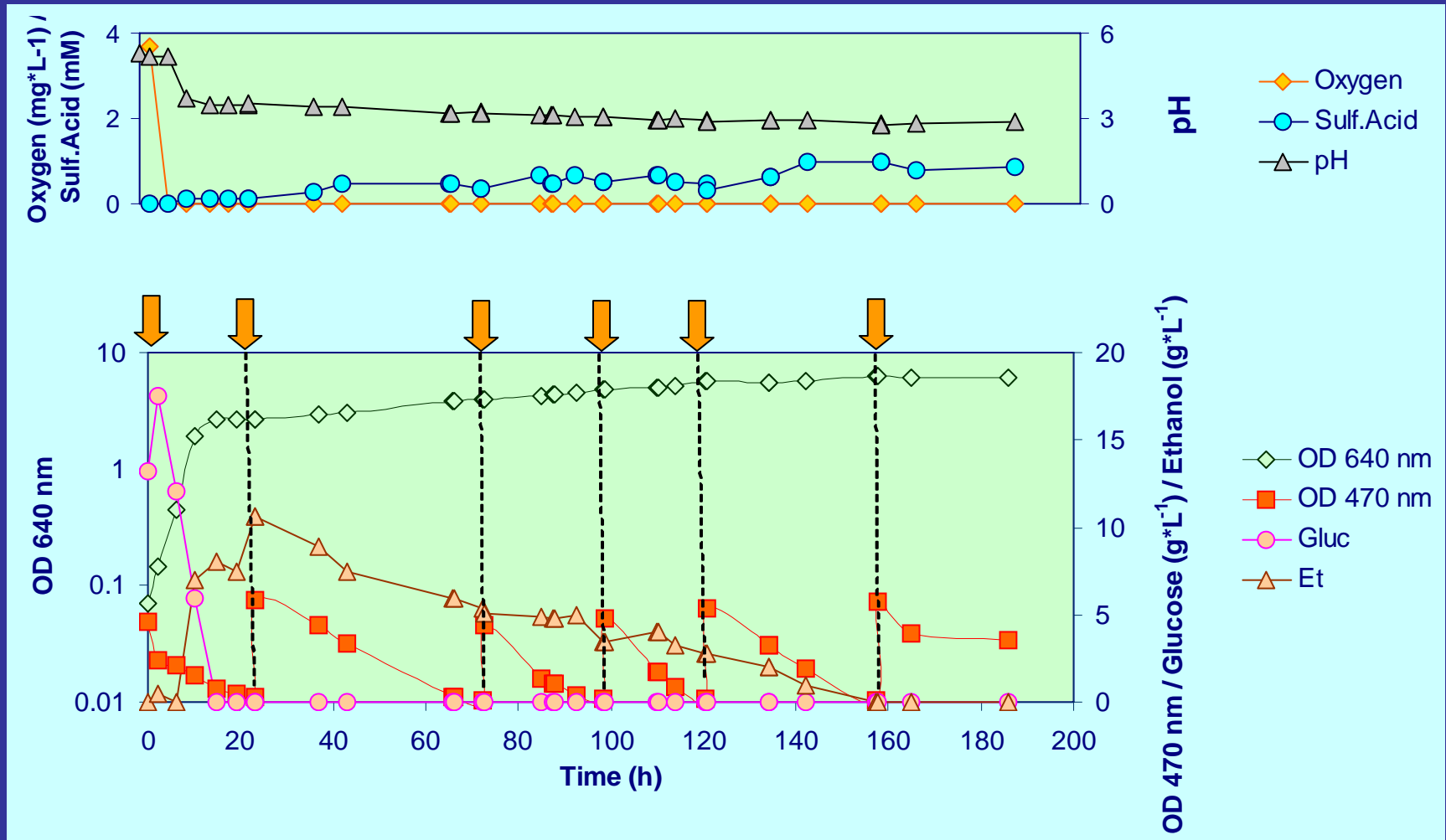
Effect of temperature



Effect of pH



The decolourisation process by yeasts



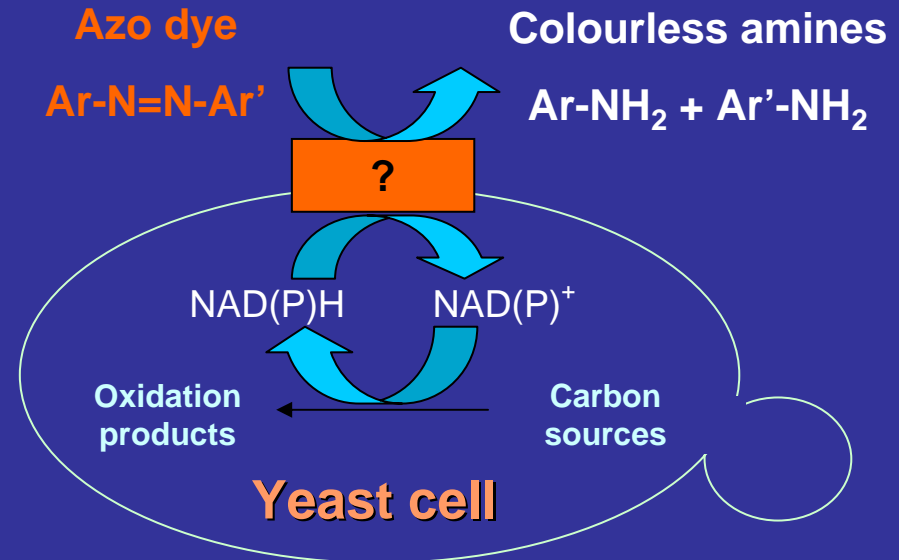
Maximum capacity of decolourisation: 0.5 mM dye/ g glucose

The decolourisation process by yeasts

Results:

- There is a need of an external carbon source;
- The activity is only detected in intact cells;
- Can use alternative carbon sources like acetate and ethanol.

Still
unanswered
question

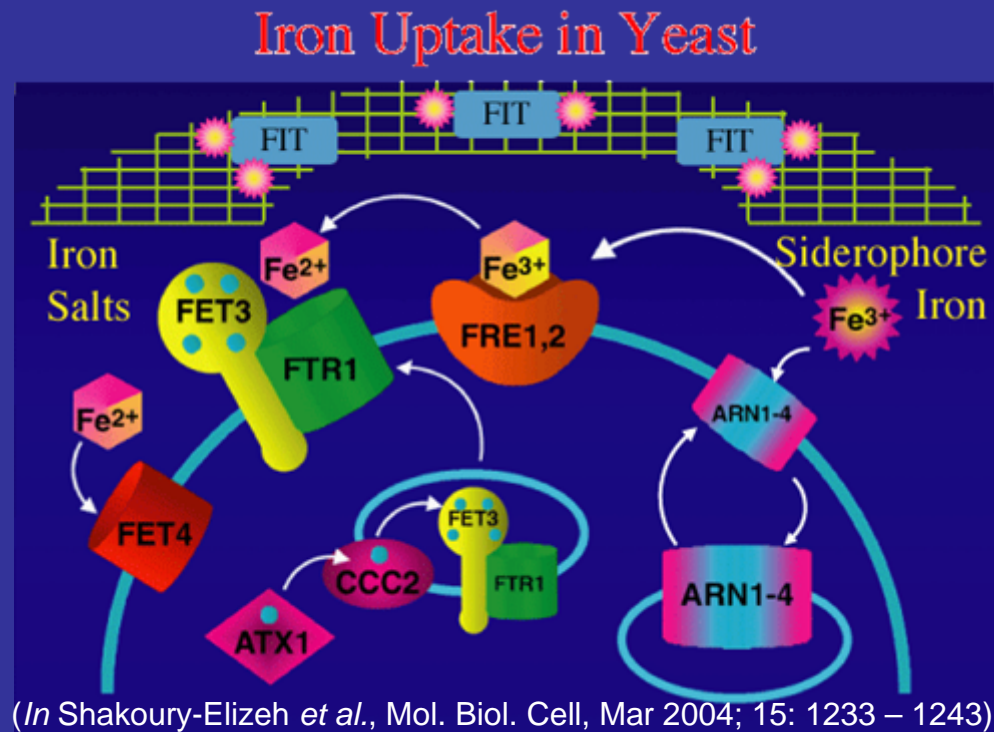


The enzyme system

From literature -> yeasts have a plasma membrane redox system to reduce iron(III) to iron (II) (solubilization of iron) – **THE FERRIC REDUCTASE SYSTEM**

Could this system be responsible for the azo reductase activity in yeast cells?

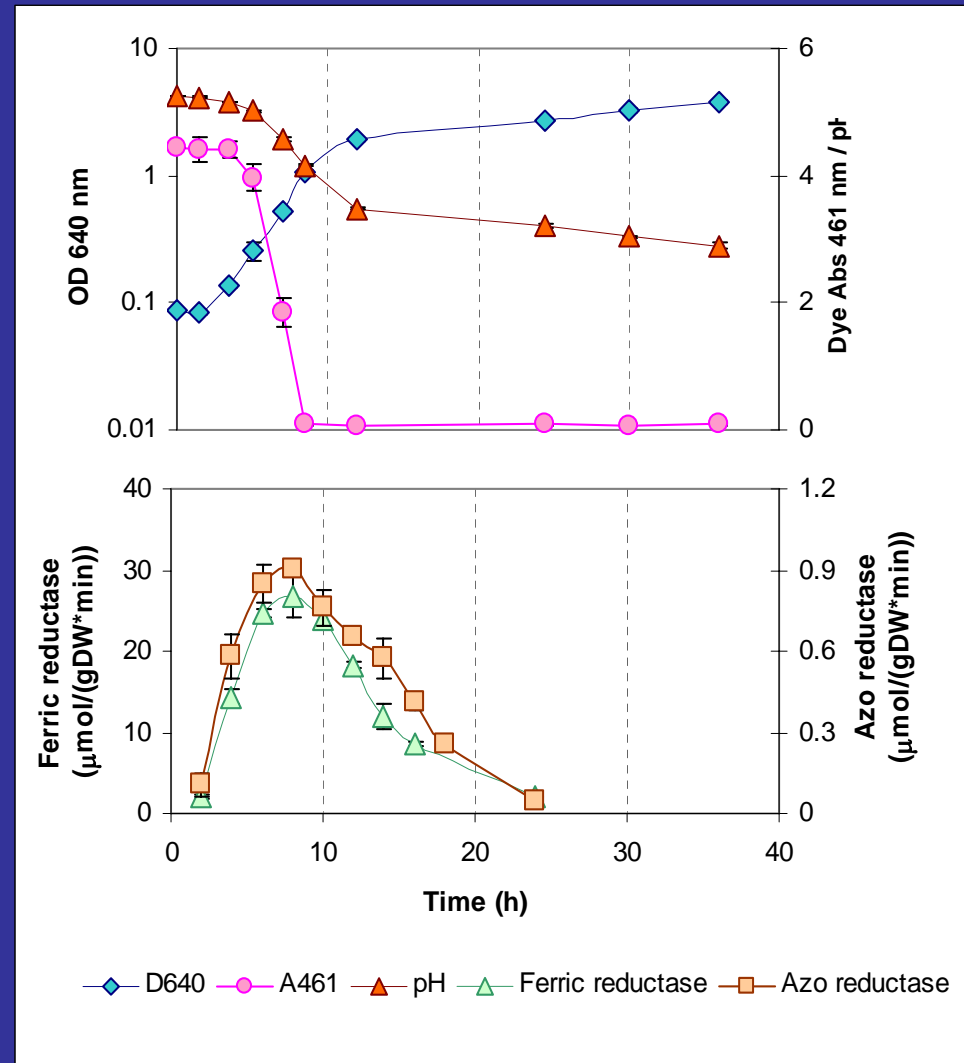
Selection of laboratorial *Saccharomyces cerevisiae* strain with the ability to degrade azo dyes



Strategies to prove the new functionality of ferric reductase

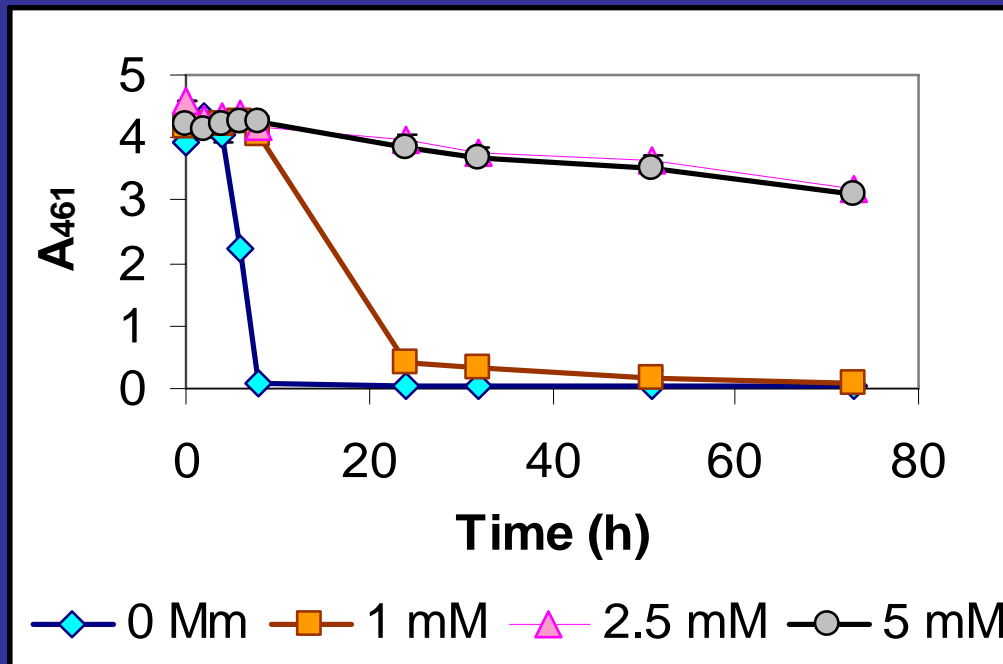
I. Measurement of both ferric and azo reductase activities along growth and decolourization:

Ferric and azo reductases have parallel activity curves with maxima in the late exponential growth phase



Strategies to prove the new functionality of ferric reductase

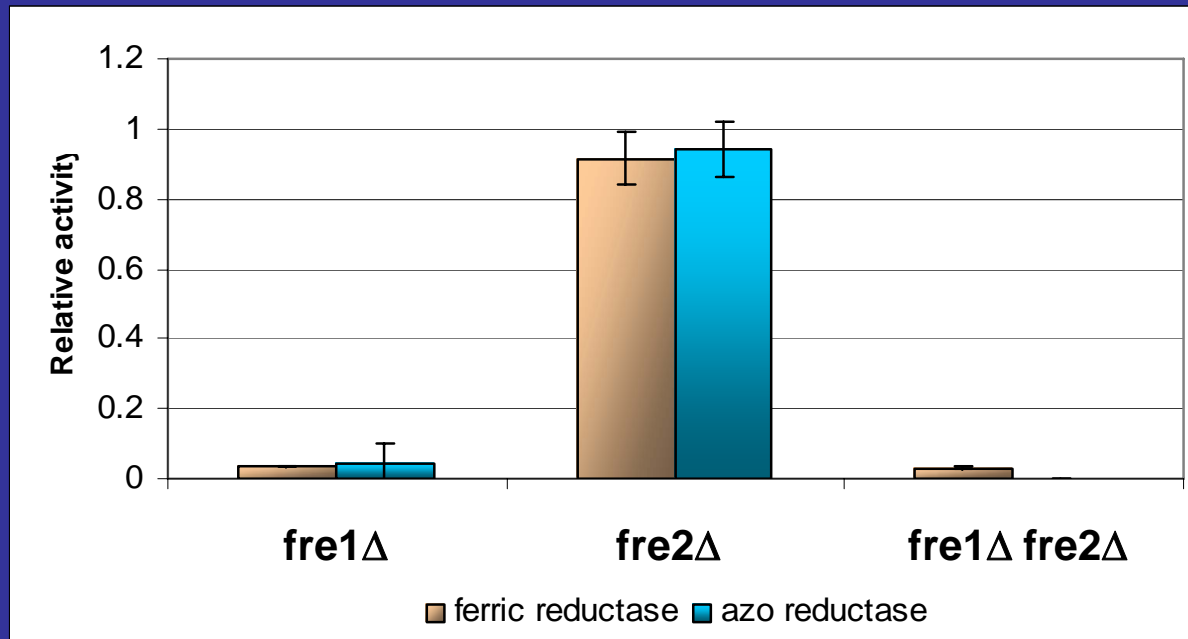
II. Addition of an known inhibitor of ferric reductase (IRON) to the decolourisation medium:



The addition of iron to the medium inhibits ferric reductase at both transcriptional and post-transcriptional levels (*Lesuisse et al. 1996*) and delays decolorization

Strategies to prove the new functionality of ferric reductase

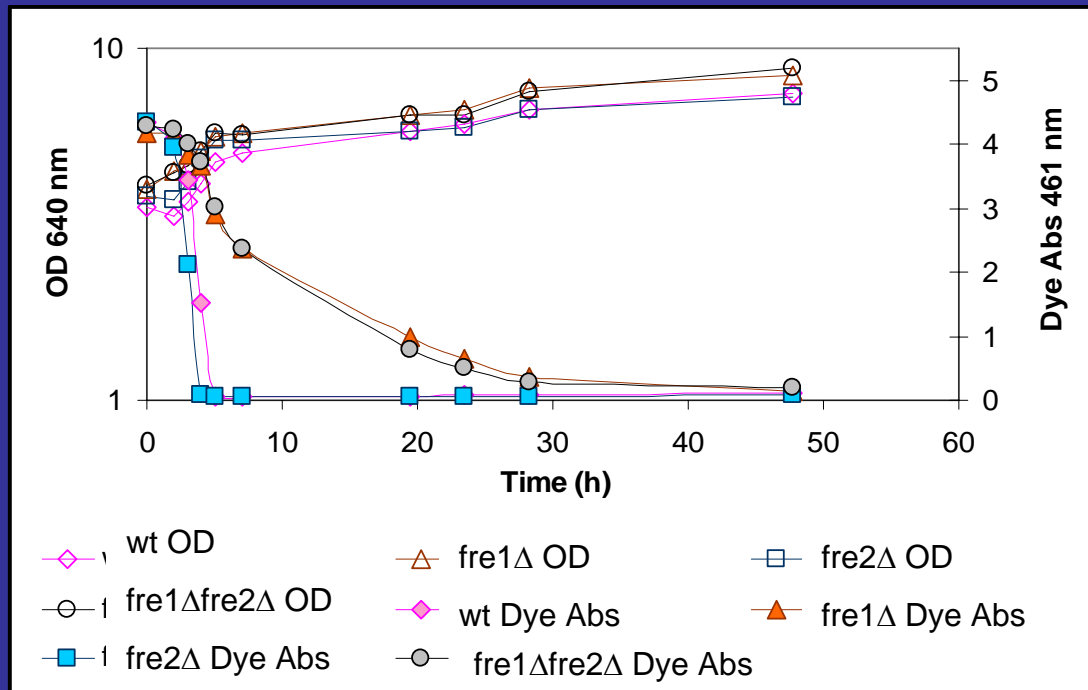
III. Deletion of the genes *FRE1* and *FRE2* in the strain: effect on azo and ferric reductase activities



The deletion of *FRE1* gene removes the decolorizing activity; *FRE2* removal does not affect the decolorizing ability, in our conditions

Strategies to prove the new functionality of ferric reductase

III. Deletion of the genes *FRE1* and *FRE2* in the strain: effect on decolourisation activity



Reduction of decolourising capacity by deletion of *FRE1* and *FRE2*

Fre1p is responsible for the major part of the azo reductase activity of intact yeast cells, but there is an alternative reductase in yeast cells

Conclusions

- Yeasts are able to reductively decolourize azo dyes in acid effluents
- It is needed an external carbon source, that can be acetate, ethanol or glucose
- Oxygen is needed in limited conditions
- The amines produced are used as carbon and nitrogen sources – the complete mineralization can be achieved
- The enzyme system responsible for the major part of this activity is the plasma membrane ferric reductase

Future perspectives

- Production of a bioreactor to study the effect on the decolourisation process of salts, pH and temperature and test real effluents
- Genetic manipulation of the *Saccharomyces cerevisiae* strain to overexpress the enzyme system

Acknowledgments

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- Thanks to all of you for your attention!